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Key indicators

Single-crystal X-ray study T = 100 KMean σ (C–C) = 0.005 Å R factor = 0.061 wR factor = 0.178 Data-to-parameter ratio = 23.2

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

Bis[3-(trimethylsilyl)propyl] methylenebisphosphonate

The title compound, also known as P,P'-di[3-(trimethylsilyl)-1propyl] methylenebisphosphonic acid, H₂DTMSP[MBP], C₁₃H₃₄O₆P₂Si₂, was crystallized from a neat oil. The compound exists in the solid state as a hydrogen-bonded infinite chain along the *a* axis. Received 25 October 2004

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Comment

Symmetrically substituted dialkyl esters of methylenebisphosphonic acid are effective solvent extraction reagents for the removal of actinide ions from acidic aqueous solutions into conventional organic solvents (Chiarizia *et al.*, 1996; Chiarizia & Herlinger, 2004). The aggregation and solvent-extraction behavior of the silyl-substituted partial esters closely mimics that of the analogous alkyl-substituted extractants, with the silyl-substituted compounds typically extracting metal ions two to three times more efficiently (McAlister *et al.*, 2001). Interestingly, the supercritical carbon dioxide (SCCO₂) solubility of H₂DTMSP[MBP] is also significantly higher than that of the di-2-ethylhexyl extractant H₂DEH[MBP], suggesting that the incorporation of silyl functionalities could be a useful method for increasing the SCCO₂ solubility of compounds (Herlinger *et al.*, 2003).



Recently, a high-yield synthetic route to symmetric dialkylsubstituted methylenebisphosphonic acids was reported (Stepinski & Herlinger, 2002). Owing to their low melting points, it has been difficult to obtain X-ray diffraction quality single crystals of these compounds, despite numerous attempts (Chiarizia *et al.*, 1996; Griffith-Dzielawa *et al.*, 2000; Stepinski & Herlinger, 2002). We report here the crystal structure of one of those esters, namely bis[3-(trimethylsilyl)propyl] methylenebisphosphonate, H₂DTMSP[MBP], (I). The compound crystallizes with a single molecule in the asymmetric unit (Fig. 1). In the solid state, the molecules are linked together by

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Figure 1

Displacement ellipsoid plot (50%) of the molecule of H₂DTMSP[MBP].

hydrogen bonds to form an infinite chain along the *a* axis (Fig. 2). Each molecule of H₂DTMSP[MBP] is bound to each of two neighboring molecules by two hydrogen bonds, with an average donor-acceptor distance of 2.506 Å (Fig. 3). In contrast, in the solution phase (toluene), H₂DTMSP[MBP] is found by vapor phase osmometry to exist primarily as a strongly hydrogen-bonded dimer (Herlinger *et al.*, 2003). Two of the possible proposed structures of this dimer (Barrans *et al.*, 1999) are shown in Fig. 4.

Experimental

The title compound was prepared according to the procedure of Stepinski & Herlinger (2002). Crystals suitable for study (m.p. 312–314 K) were obtained by slow growth at room temperature from a H₂DTMSP[MBP] sample initially isolated as an oil. Melting point, equivalent weight by titration, ¹H NMR spectra, and ³¹P NMR spectra are consistent with those previously reported (Stepinski & Herlinger, 2002; McAlister *et al.*, 2002).

Crystal data

$C_{13}H_{34}O_6P_2Si_2$	Z
$M_r = 404.52$	D
Triclinic, $P\overline{1}$	Μ
a = 6.534 (2) Å	C
b = 11.130 (2) Å	
c = 16.325 (2) Å	θ
$\alpha = 100.64 \ (1)^{\circ}$	μ
$\beta = 100.25 \ (1)^{\circ}$	Т
$\gamma = 106.23 \ (2)^{\circ}$	Pı
$V = 1086.6 (4) \text{ Å}^3$	0.

Data collection

Oxford Diffraction Xcalibur3 diffractometer equipped with a CCD detector φ and ω scans Absorption correction: numerical (Clark & Reid, 1995) $T_{min} = 0.888, T_{max} = 0.983$ 10 850 measured reflections

Refinement

Refinement on F^2 $R[F^2 > 2\sigma(F^2)] = 0.061$ $wR(F^2) = 0.178$ S = 1.154971 reflections 214 parameters Z = 2 $D_x = 1.236 \text{ Mg m}^{-3}$ Mo K\alpha radiation Cell parameters from 3959 reflections $\theta = 1.8-29.4^{\circ}$ $\mu = 0.33 \text{ mm}^{-1}$ T = 100 (2) KPrism, colorless $0.36 \times 0.13 \times 0.02 \text{ mm}$

4971 independent reflections 3556 reflections with $I > 2\sigma(I)$ $R_{int} = 0.046$ $\theta_{max} = 27.5^{\circ}$ $h = -8 \rightarrow 7$ $k = -14 \rightarrow 14$ $I = -21 \rightarrow 21$

H-atom parameters constrained $w = 1/[\sigma^2(F_o^2) + (0.1P)^2]$ where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{max} < 0.001$ $\Delta\rho_{max} = 1.00 \text{ e } \text{Å}^{-3}$ $\Delta\rho_{min} = -0.57 \text{ e } \text{Å}^{-3}$







Figure 3

View of the extended structure of $H_2DTMSP[MBP]$. Each molecule is bound by four hydrogen bonds to two adjacent $H_2DTMSP[MBP]$ molecules to form an infinite chain along the *a* axis. The bulk of the 3-(trimethylsilyl)propyl groups have been removed for clarity. Hydrogen bonds are shown as dashed lines. The view is approximately along [011].



Figure 4 The proposed dimeric structures of H₂DTMSP[MBP] in toluene.

Table 1	
Selected geometric par	rameters (Å, °).

O1-P1	1.475 (2)	O4-P2	1.481 (2)
O2-P1	1.539 (2)	O5-P2	1.543 (2)
O3-P1	1.572 (2)	O6-P2	1.559 (2)
O1-P1-O2	115.79 (12)	O4-P2-O5	113.86 (12)
O1-P1-O3	113.16 (12)	O4-P2-O6	116.29 (12)
O2-P1-O3	100.31 (11)	O5-P2-O6	104.70 (11)
O1-P1-C13	111.17 (13)	O4-P2-C13	111.41 (12)
O2-P1-C13	109.79 (12)	O5-P2-C13	108.38 (13)
O3-P1-C13	105.72 (12)	O6-P2-C13	101.16 (13)

Table 2

Hydrogen-bonding geometry (Å, °).

$D - H \cdots A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - H \cdot \cdot \cdot A$
$\begin{array}{c} O2{-}H2{\cdots}O4^{i}\\ O5{-}H5{\cdots}O1^{ii} \end{array}$	0.82	1.78	2.518 (3)	148
	0.82	1.83	2.493 (3)	137

Symmetry codes: (i) -x, 1 - y, -z; (ii) 1 - x, 1 - y, -z.

H atoms were positioned geometrically (C–H = 0.96–0.97 Å; O– H = 0.82 Å) and refined using a riding model, with fixed individual displacement parameters $[U_{iso}(H) = 1.2U_{eq}(C)$ for methylene H, $1.5U_{eq}(C)$ for methyl H or $1.5U_{eq}(O)$].

Data collection: *CrysAlisCCD* (Oxford Diffraction, 2004); cell refinement: *CrysAlisRED* (Oxford Diffraction, 2004); data reduction: *CrysAlisRED*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97*

(Sheldrick, 1997); molecular graphics: *ORTEP-3 for Windows* (Farrugia, 1997); software used to prepare material for publication: *WinGX* (Farrugia, 1999).

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